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Gas (# of moles of gas is constant) Gas # of moles of Gas in container (mol) Volume of Container (L) Temperature of Gas (K) Pressure of Gas (kPa) A 0.3008 14.0 672.0 120.0 B 0.3008 16.0 1280 200.0 C 0.3008 10.0 340.0 85.0 Table 2: Experimental Data for Three Different Samples of Gases That Contain Different Amounts of Gas (# of moles of gas is ...

Information: The Gas Constant

Chem 1A-Lec 12-Moles Part 1 CHEMQUEST 02 [CBSE AIPMT 2007] Concentrated aqueous sulphuric acid is 98% H₂SO₄ by mass and has a density of 1.80 gmL⁻¹. Volume of CHEMQUEST 01 A mixture of two gases A and B in the mole ratio 2:3 is kept in a 2 litre vissel.a second 3 litre vessel has the same two gases in the

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3. On planet X, 2.78 moles of a gas takes up 1.85 L under a pressure of 74.1 kPa and a temperature of 201oC. What is the value of the ideal gas constant (R) on planet X? (include units) 0.104 kPa-L/(mol-K) 4. At a pressure of 103 kPa and a temperature of 22oC, 52.9 g of a certain gas has a volume of 31.5 L. What is the identity of this gas?

Skill Practice 36 - Roosevelt High School AP Chemistry ...

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Skill Practice 36 a) How many moles of gas are in his lungs? 0.343 mol b) If the diver swims to the surface without exhaling where the temperature is 32°C and the pressure changes to 100.2 kPa, what will the volume of the air in his lungs be? 868 L 3 On planet X, 278 ... Yield 103 ChemQuest #35: Intro to Gases 105 ChemQuest #36: Gases and Moles 109

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contains 3 moles of hydrogen gas and 2 moles of oxygen gas The total pressure in the container is Chemquest 36 Answers LTHS Answers Skill Practice 36 Name: ____ Date: ____ Hour: ____ 1 Under water where the temperature is 17°C and the pressure is 394 kPa, a diver inhales 21 L of air

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