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Atoms: The Building

Blocks of Matter

SECTION 3 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

Explain the difference

between the mass

number and the atomic

number of a nuclide.

Mass number is the

total number of

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protons and neutrons
in the nucleus of an
isotope.

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Gases Class SHORT

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ANSWER Answer the following questions in the space provided. c c
The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol. nRT (c) multiplied by 22.4 L. (d) divided by 22.4 L.
For the expression $V =$ (a) increasing P

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...

3 react according to
the above equation,
how many moles of NO
will be consumed? 8.

Propyne gas can be

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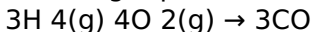
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used as a fuel. The combustion reaction of propyne can be represented by the

following equation: C



a. Write all the possible mole ratios in this system. 4 mol O₂:1 mol C₃H₄; 3 mol CO₂:1 mol C₃H₄; 2 ...

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REVIEW The Periodic
Law SECTION 1 SHORT

ANSWER Answer the
following questions in
the space provided. 1.

c In the modern
periodic table,
elements are ordered
(a) according to
decreasing atomic
mass.

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Chapter 5 Section 3

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CHAPTER 3 REVIEW

Atoms: The Building

Blocks of Matter

SECTION 3 SHORT

ANSWER Answer the

following questions in
the space provided. 1.

Explain the difference
between the mass
number and the atomic
number of a nuclide.

Mass number is the
total number of
protons and neutrons
in the nucleus of an

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isotope.

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Chapter 2 Section 3

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SECTION 3 continued

5. Explain the following properties of solids by describing what is occurring at the atomic level. a. Metallic solids conduct electricity well, but covalent network solids do not. Metals have many electrons that are not bound to any one atom;

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therefore they are able to move throughout the crystal.

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Name Section Quiz,

continued Class Date

5. In some instances, the concentration of a solution is expressed as molality instead of molarity because a. molality is easier to calculate. b. molarity

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Chemical Bonding

CHAPTER 6 STUDY

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IONIC BONDING AND

IONIC COMPOUNDS

SHORT ANSWER

Answer the following questions in the space provided. 1. _____ The notation for sodium

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chloride, NaCl, stands for one (a) formula unit.

Modern Chemistry Chapter 6 Review Answers Section 3

CHAPTER 3 REVIEW

Atoms: The Building Blocks of Matter

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic

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3 Review Answers

number of a nuclide.

Mass number is the total number of protons and neutrons in the nucleus of an isotope.

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