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Numericals Chemistry Chapter Solid State

Ans: In solid state, the ions cannot move, they are held by strong electrostatic forces of attraction. So, ionic solids do not conduct electricity in solid state. However, in the molten state, they dissociate to give free ions and hence conduct electricity. 1.9. What type of solids are electrical conductors, malleable and ductile?

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Lets see "How to solve Numericals of class 12th chemistry chapter Solid state". In this chapter numericals are mainly of 4 types -. (a) Calculation of Rank of the unit cell - Rank of the unit cell means the number of atoms actually present in a unit cell.

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Chemistry Notes for class 12 Chapter 1 The Solid State Solids Solids are the chemical substances which are characterised by definite shape and volume, rigidity, high density, low compressibility. The constituent particles (atoms, molecules or ions) are closely packed and held together by strong interparticle forces Types of Solids

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The Solid State Class 12 Notes Chapter 1. 1. Solids. Solids have definite volume, shape, and mass due to the short distance between the fixed position of particles and strong interactions between them.

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