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Stoichiometric Calculations are Based on Chemical Formulas
Let's learn some terms used in Stoichiometry first. Formula Mass: It is the sum of the atomic weights of the various atoms present in the molecule of the substance. For example, we can calculate the formula mass of Na_2S as $2(23) + 1(32) = 78$

Stoichiometry and Stoichiometric Calculations: Concepts

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Stoichiometry Stoichiometric Calculations Starting with 10. g of $\text{C}_6\text{H}_{12}\text{O}_6$... we calculate the moles of $\text{C}_6\text{H}_{12}\text{O}_6$... use the coefficients to find the moles of H_2O & CO_2 and then turn the

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moles to grams $C_6H_{12}O_6(s) + 6 O_2(g) \rightarrow 6 CO_2(g) + 6 H_2O(l)$
10.g ? + ? Example: 10 grams of glucose ($C_6H_{12}O_6$) react in a combustion reaction.

Chapter 3 Stoichiometry - Chemistry

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A stoichiometric diagram of the combustion reaction of methane. Stoichiometry / ,stɔɪki'ɒmɪtri / is the calculation of reactants and products in chemical reactions in chemistry.

Stoichiometry - Wikipedia

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The text presents the fundamentals of chemical engineering operations and processes in a simple style that helps the

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students to gain a thorough understanding of chemical process calculations. The...

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