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An atom is the smallest unit of an element that maintains the chemical identity of that element. An element is a pure substance that cannot be broken down into simpler, stable substances and is made of one type of atom.

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1. each element is made up of particles called atoms
2. all atoms of an element are identical in size, mass, etc. (incorrect --> ions)
3. atoms cannot be divided, created, or destroyed (incorrect --> atomic bomb used divided atoms)
4. atoms of different elements combine in whole-number ratios to form compounds

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Molecules: the smallest unit of a substance that keeps all of the physical and chemical properties of that substance; it can consist of one atom or two or more atoms bounded together. (usually consists of two or more atoms combined in a definite ratio).

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Wednesday: Complete FRONT (1-4) of density worksheet AND finish your metric olympics lab
Tuesday: Study for the elements quiz and start working on the Metrics Olympics lab (due Thursday at the beginning of class)!
Monday: Rounding Sig.Figs with math (1-13 on front of notes wkst) Here is a video to help: Last Week

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6 Study Guide for An Introduction to Chemistry Problems Key Section 1.4 Measurement and Units 15. Complete

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the following table by writing the property being measured (mass, length, volume, or temperature) and either the name of the unit or its abbreviation. (Objs 3 and 7)

Chapter 1 An Introduction to Chemistry

Chemistry Formative Unit 1. Chemistry. Big Bang Theory. Universe. Matter. Chemistry is the study of the universe, matter and the changes.... Some evidence suggests that the Universe started with a big ex.... Created during the Big Bang and consists of all matter, energy.... Any substance with mass and volume.

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chemistry through the use of the following items: Reading assignments; Activities related to the reading

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AP Chemistry Study Strategies. Before we get to notes on content, here are some study tips that you should keep in mind as you review. In a subject like chemistry, there's a huge difference between looking over the material and actually learning it. #1: Start With the Basics. AP Chemistry is a subject that builds on itself from the ground up.

The Ultimate AP Chemistry Study Guide - PrepScholar

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Unit 1 - Foundations of Chemistry

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Unit 1 - Chemistry of Life - MY SITE - HOME

Unit 1 Chemistry Study Guide What characterizes a prokaryotic cell? Why do experiments need a control? Why do scientists use the scientific method to study environmental problems? List the hierarchy of organization for living things & for classification.

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